The 4<sup>th</sup> International scientific Conference on Nanotechnology& Advanced Materials & Their Applications (ICNAMA 2013)3-4 Nov, 2013

# Structural and Optoelectronical Properties of $In_2S_3$ Thin Films Prepared by CSP Technique for Solar Cell Application

Dr. Baha. T. Chiad Science college, University of Baghdad /Baghdad M. Sh. Essa Ministry of Science and Technology / Baghdad M. D. Abd-aljabar Ministry of Science and Technology / Baghdad Email: mohalnasban@yahoo.com (M.Sh. Essa). J. A. Abd-aula Ministry of Science and Technology / Baghdad

# ABSTRACT

Indium Sulfide  $In_2S_3$  window layer have been prepared by Chemical Spray Pyrolysis (CSP) at substrate temperate Equal (573 K) from Indium chloride and Thiourea were In/S ratio equal 1.2/8 in the spray solution, the samples prepared with different thicknesses (1.6, 1.7, 2.0 µm), the structural, optical and electrical of these films was investigated at different annealing temperature (Ta).X-ray diffraction studied shows the Structural properties of this layer are polycrystalline with preferred orientation 221, and have good improvement in the crystal structure at the annealing temperature (573K for 1h). The grain size increase with increasing annealing temperature and the optical band gap was found in the range (2.4-2.55 eV) as a function of the film thicknesses and the annealing temperature. Electrical studied of the sprayed and annealed sample shows n-type electrical conductivity, the mobility improved at the annealing temperature equal (573 K) but the resistivity decreased with this temperature.

**Keywords:** In<sub>2</sub>S<sub>3</sub> Window Layer; Chemical Spray Pyrolysis; Structural; Optical; Electrical.

الخصائص التركيبيه البصريه والكهربانيه لاغشية كبريتيد الانديوم In<sub>2</sub>S<sub>3</sub> الرقيقه المحضرة بطريقة الرش الكيميائى الحراري المستخدمه فى تطبيقات الخلايا الشمسيه

الخلاصة:

تم تحضير طبقة نفوذ من مادة كبريتيد الانديوم بطريقة الرش الكيميائي الحراري على ارضيات زجاجيه عند درجة حرارة قدرها 573 كلفن من مادة كلورايد الانديوم والثايوريا بنسبة انديوم/

629

كبريت يساوي 8/1.2 في محلول الرش . حضرت النماذج باسماك مختلفة (1.6 , 1.7 , 2) مايكرومتر . تم دراسته الخصائص التركيبية , البصرية والكهربائيه لهذه الاغشيه عند درجات حرارة تلدين مختلفه. اظهرت نتائج حيود الاشعة السينيه ان هذه الطبقه هي ذات تركيب بلوري متعدد التبلور عند الاتجاه السائد 221 لمعاملات ميلر ,وكذلك تحسن جيد في البنية البلوريه عند درجة حرارة التلدين ( 573 كلفن ولمدة ساعة واحده ) .كما وان الحجم الحبيبي ازداد بزيادة درجة حرارة التلدين . بينت الفحوصات البصريه ان فجوة الطاقة البصرية تغيرت بالمدى ( 2.4 – 2.5 الكترون فولت) كدالة لدرجة حرارة التلدين والتحضير . اظهرت نتائج الدراسة الكهربائية للنماذج عند درجة حرارة التحير والمدنه بدرجات مراريه منتخذ انها من نوع حاملات الشحنة ذات التوصيلية السالبة , كما وان قيم التحركية تحسنت ولكن قيم المقاومية قلت عند درجة حرارة التلدين 573 كلفن .

# **INTRODUCTION**

ecently there has been increasing interest in research on III-VI materials because these material have found great use in the electronic industry, Indium sulfide  $(In_2S_3)$  is a promising compound which is used in optoelectronic or photovoltaic application [1-3], The In<sub>2</sub>S<sub>3</sub> compound has a potential application in photovoltaic devices these structures used as window layer [4,5], due to the wide energy gap [6], and photosensitivity [7].which could be prepared using different methods such as thermal method [8] ,radio-frequency (rf) sputtering [9],atomic evaporation laver epitaxy[10], Chemical bath deposition [11] Spray Pyrolysis (SP) is one of the low cost techniques widely employed to deposit indium sulfide thin films [12,13], this technique has the advantage of being cheap and simple.  $In_2S_3$  thin films appear to be a promising candidate for photovoltaic applications. It can be used as an affective nontoxic cadmium sulfide (CdS) in CIS based Solar cells.

The motivation behind this is not only to eliminate toxic cadmium but also to improve light transmission in the blue wavelength region by using a material having band gap wider than of CdS .CuInS<sub>2</sub>/In<sub>2</sub>S<sub>3</sub> based solar cell with In<sub>2</sub>S<sub>3</sub> as the buffer layer could reach efficiencies (12.4%) [14] The primary function of a window layer in hetrojunction is to form a junction with the absorber layer while admitting a maximum amount of light to the junction region and absorber layer, no photocurrent generation occurs in the window layer For high optical throughput with minimal resistive loss the band gap of the window layer should be as high as possible and as thin as possible to maintain low series resistance.

#### **EXPERIMENTAL**

Indium sulfide  $In_2S_3$  as n-type layer can be used as an optical window in photovoltaic cells and it can constitute a good alternative to CdS layers[15]. This layer have been prepared by CSP technique, which is simple and low coast technique, using Indium chloride (InCl<sub>3</sub>) and Thiourea (CS(NH<sub>2</sub>)<sub>2</sub>) in the spray solution with constant molar concentration (0.1 M) sprayed on glass substrate in the dimension (25mm×25mm). Total volume of solution sprayed for depositing was 50 ml with In/S ratio 1.2/8 in the films. In the normal condition, using nitrogen (N<sub>2</sub>) as carrier gas the distance between the nozzle and the substrate is 30 cm and 1 par is the pressure of the carrier gas. The preparation temperature was 573 K and using different annealing temperature (T<sub>a</sub>) (523,573,623 K) for one hour (1 h). In the present work, spray solution of indium chloride and Thiourea shows here that possible to prepare  $In_2S_3$  thin films using CSP Technique, these films have obtained shows homogeneous and adherent. Thin films of  $In_2S_3$  were prepared having different thicknesses in (µm) dimension. Structural of  $In_2S_3$  thin films as prepared and annealed samples were determined using X-ray diffraction type Shimadzue (XRD6000) diffractometer have Cuk $\alpha$  radiation with  $\lambda$ = 1.5406 A<sup>0</sup>. Optical properties were studied using optical absorbance and transmittance spectrum (UV VIS NIR) spectrophotometer Shimadzue (UV/1650PC), and electrical properties have measured used (ECOPIA HMS-3000) Hall Effect measurement System.

# **RESULT AND DISCUSSION** Optical analysis

Figure (1) shows the transmission spectrum T% of the as- deposited (sprayed)  $In_2S_3$  films at (573 K), Figure (2) shows the films annealed at (523, 573, 623 K) at different thickness(1.6, 1.7,2µm) with sample name (S1,S2,S3) respectively from this Figures (1,2) it can be seen that T% is slightly increased with decreasing thickness. Transmission spectra recorded in the wave length rang (300- 900) nm. In order to determine the optical band gap,  $(\alpha hv)^2$  versus hv graph was plotted, which is found in the range (2.46- 2.54) eV depending on film thickness. For these samples it's clearly energy gap shift towards lower values as thickness increased. Thickness dependence of band gap could arise due to one or combined effect of the following cause along density of dislocations quantum size effect and the change in barrier height due to change in grain size in polycrystalline films. Band gap variation of all the sample in as- sprayed and different annealed temperature is given in Figures. (3), (4) respectively.

$$\alpha(hv) = A(hv - E_g)^{1/2}$$
[16]

Where A is a constant and Eg is the corresponding semiconductor band gap  $\alpha$  is absorption coefficient. This result having significance in solar cells applications as buffer or window layer. Wider band gap materials improve the light transmission in blue region resulting in increase of the short circuit current and open circuit voltage in solar cells.



Figure (1) Shows transmission spectrum of In<sub>2</sub>S<sub>3</sub> As-sprayed.

Eng. &Tech.Journal, Vol. 32,Part (B), No.4, 2014

Structural and Optoelectronical Properties of *In*<sub>2</sub>*S*<sub>3</sub> Thin Films Prepared by CSP Technique for Solar Cell Application







Figure (3)  $(ahv)^2$  versus the photon energy (hv) of  $In_2S_3$  layer as-sprayed at different thicknesses.



Figure (4)  $(ahv)^2$  versus the photon energy (hv) of  $In_2S_3$  layer with different ( $T_a$ ).

#### **Structural Analysis**

The structural analysis of indium sulfide thin films was carried out using X-ray diffractometer XRD with varying diffraction angle (2 $\theta$ ) from 20<sup>0</sup> – 60<sup>0</sup>.

Figure (5) shows the XRD pattern of indium sulfide thin film at fixed ratio [In]/[S] equal 1.2 / 8 with different annealing temperature (T<sub>a</sub>) (523, 573,and 623 K). These films showed polycrystalline structure according to the standard card (JCPDS 25-0390) with tetragonal structure, the strongest peak of the film have preferred orientation along (221) and (109), these highest intensity peak describes good crystallinity material prepared by this technique. the other low intensity peaks correspond to the (138), (401) plane. The crystallographic parameter of the film was collected and computed with different annealing temperature T<sub>a</sub> and presented in Table (1).

The crystallite size of  $In_2S_3$  thin film was calculated using the Scherer's formula. Lattice parameter computed and *d*-value computed and incurred from XRD data observed and compared with (JCPDS 25-0390) and found good agreement with standard card.. Moreover the variation in grain size with different Ta, as shown in Figure (6). from this figure, it is clear that, the grain size increased with increasing annealing temperature.



Figure (5). XRD pattern for In<sub>2</sub>S<sub>3</sub> layer at different T<sub>a</sub>.



Figure (6). Variation in grain size of In<sub>2</sub>S<sub>3</sub> at different T<sub>a</sub>.

 Table (1) Shows lattice parameter and d-value observed compared with standard card of the In<sub>2</sub>S<sub>3</sub> layer.

| Temp<br>K  | 20     | hkl | I/Io | FWHM   | a<br>observed | a<br>standard | c<br>observed | c<br>standard | d<br>observed | d<br>standard |      |
|------------|--------|-----|------|--------|---------------|---------------|---------------|---------------|---------------|---------------|------|
| As-<br>dep | 33.483 | 221 | 100  | 0.6698 | 7.5893        | 7.619         | 32.3392       | 32.329        | 2.6742        | 2.6844        | 0.38 |
|            | 27.604 | 109 | 97   | 0.815  |               |               |               |               | 3.2288        | 3.2327        | 0.12 |
|            | 47.995 | 401 | 54   | 0.6517 |               |               |               |               | 1.894         | 1.9015        | 0.39 |
| 523        | 33.474 | 221 | 100  | 0.5228 | 7.5913        |               | 32.3252       |               | 2.6749        | 2.6844        | 0.36 |
|            | 27.609 | 109 | 88   | 0.6678 |               |               |               |               | 3.2283        | 3.2327        | 0.14 |
|            | 47.981 | 401 | 47   | 0.6045 |               |               |               |               | 1.8946        | 1.9015        | 0.36 |
| 573        | 33.415 | 221 | 100  | 0.4729 | 7.6034        |               | 32.4269       |               | 2.6795        | 2.6844        | 0.19 |
|            | 27.537 | 109 | 80   | 0.6164 |               |               |               |               | 3.2366        | 3.2327        | 0.12 |
|            | 47.901 | 401 | 36   | 0.5505 |               |               |               |               | 1.8975        | 1.9015        | 0.21 |
| 623        | 33.421 | 221 | 100  | 0.4138 | 7.6009        |               | 32.3966       |               | 2.679         | 2.6844        | 0.2  |
|            | 27.557 | 109 | 90   | 0.5205 |               |               |               |               | 3.2343        | 3.2327        | 0.05 |
|            | 47.920 | 401 | 40   | 0.4925 |               |               |               |               | 1.8968        | 1.9015        | 0.24 |

# In<sub>2</sub>S<sub>3</sub> Layer (n-Type) Electrical Properties

Electrical resistivity, mobility, carrier concentration and also that conductivity were done using Hall Effect system at (R.T) for In/S ratio equal 1.2/8 in the solution. Indicated from that the film was n- type and that due to "S-rich" in the films comparable with "Inpoor" that lead to n-type formation, We can see that the result of carrier concentration is 5.731E+11 ,the mobility is 1.161E+3, the resistivity is 9.385E+3,and the conductivity is 1.065E-4 this result match well with the result of Ratheesh et.al .[85] from Figure (7) we can see, that the I-V curve for n-type layer collected from hall effect system.

Eng. &Tech.Journal, Vol. 32,Part (B), No.4, 2014



Figure (7) Shows the I-V curve for  $In_2S_3$  (n-type) layer.

# CONCLUSIONS

We reported on the deposition of  $In_2S_3$  by the spray pyrolysis technique using indium chloride and thiouea in the ratio[In/S] equal to [1.2/8], XRD studies showed that the deposited materials are stoichiometric material and have good crystal structure at annealing temperature  $T_a$  573 K with preferred orientation along (109) and (221).optical studied showed this film have a direct band gap in the range (2.46,2.55)eV according to the different annealing temperatures, electrical studied showed n-type layer

# Acknowledgment

We thank and acknowledge to the Solar Energy Research Center /Renewable Energy Directorate in Ministry of Science and Technology.

#### REFERENCES

- [1]. Mathew. M. C. Sudha Kartha.K.P.Vijayakumar, "Mater Sci"20(2008)217.
- [2]. Calixto-Rodriguez, M. H. Martínez , A.Sanchez-Juarez . Thin Solid Films 517 (2009) 2332–2334.
- [3]. Indra Puspitasari, T.P. Gujar, Kwang-Deog Jung, Oh-Shim Joo, journal of materials processing technology 201(2008) 775–779.
- [4]. Yousfi, E.B. T. Asikainen, V. Pietu, P. Cowache, M. Powalla, D. Lincot, Thin Solid Films361-362 (2000) 183.
- [5]. Strohm, A. L. Eisenmann, R.K. Gebhardt, A. Harding, T. Schlotzer, D. Abou-Ras, H.W.Schock, Thin Solid Films 480-481 (2005) 162-167.

[6]. Herrero, J. J. Ortega, Sol. Energy Mater 17 (1988) 357.

[7]. John, T.T. C.S. Kartha, K.P. Vijayakumar, T. Abe, Y.Kashiwaba, Appl. Phys. A 82(2006) 703.

[8]. Timoumi, A. H.Bouzouita, M.Kanzari, B.Rezig. Thin Solid Films 480-481(2005) 124-128.

- [9]. George, J,Joseph K S,Pradeep B and Palson T I 1988 phys.Status Solidi a 106 123
- [10]. Bouguila, N. H. Bouzouita, E, Lacaze, A. Belhadj Amara, H. Bouchriha , A. Dhouib, J. Physique III 7(1997)1647
- [11]. Indra Puspitasari, T.p Guiar, Kwang-Deog Jung, Oh-Shim Joo.201(2008)775-779.
- [12]. Calixto-Rodriguez, M. A. Tiburcio-Silver, A. Ortiz, A. Sanchez-Juarez, Thin SolidFilms 480-481 (2005) 133-137.
- [13]. John, T.T. S. Bini, Y. Kashiwaba, T. Abe, Y. Yasuhiro, C.S. Kartha, K.P. Vijayakumar, Semicond. Sci. Technol. 18 (2003) 491
- [14]. Buecheler a,, S. D. Corica a, D. Guettler a, A. Chirila a, R. Verma a, U. Müller b, T.P. Niesen c, J. Palmc, A.N. Tiwari a, Thin Solid Films 517 (2009) 2312–2315
- [15]. Belgacem, S. M. Amlouk, R. Bennaceur, Rev. Phys. Appl. 25 (1990)1213.
- [16]. Hara, K. K. Sayama, H. Arakawa, Sol. Energy Mater. Sol. Cells 62(2000) 441.