A Study on the Structural, Optical and Electrical Properties of Tungsten TrioxideWO₃Thin Film for Gas Sensing Applications

Dr. Adawiya J. Haider Applied Sciences Department, University of Technology /Baghdad. Huda A. Mutasher Applied Sciences Department, University of Technology /Baghdad. Email::hudaahmed1980@gmail.com

Received on:10/7/2014 & Accepted on:30/8/2015

ABSTRACT

The pulsed laser deposition method, PLD, has been used in this project to deposit the pure Wo_3 The aim was to study the influence of temperature on surface morphology of the deposits materials using Atomic Force Microscope measurements at different temperatures of (RT,300,400)°C. The structural characteristics of the films prepared on glass substrates have been studied by using X-ray diffraction and AFM. These tests show that there is a direct relationship between the grain sizes of the nanoparticles observed at the surface and influence of temperature which means as temperature increases there will be a similar enhancement in the grain size as well. Additionally, we were able to find that the best of temperature was 400°C. Furthermore, the characterizations of the gas sensing of these thin films were strongly influenced by the surface morphology. It has been also found that nanocrystalline Wo_3 gas sensing material was presented a better sensitivity as temperature increases

Keywords: Pulsed Laser Deposition (PLD), thin films, nanocrystalline.

الخلاصة

في هذا العمل استخدمت طريقه الترسيب بلليزر النبضي لترسب ثلاثي اوكسيد التنكستن وتهدف لدراسه تأثير درجه الحراره الترسيب على مورفولوجيه السطح للمواد المترسبه باستعمال المجهر القوة الذرية و النتائج بينت ان هنالك علاقه مباشر بين الحجم الحبيبي للتراكيب الملاحظه عند السطح وتأثر درجه الحرارة وهذا يعني عند زيادة درجه الحراره يزداد الحجم الحبيبي وكانت افضل درجه حراره هي 400 سليزي . كما ان مورفولوجه السطح لها تأثير على الخصائص التحسسيه لهذه الاغشية وأيضا وجدالتحسسيه ثلاثي اوكسيد التنكستن تزداد مع زياده درجه الحراره .

1473

https://doi.org/10.30684/etj.2015.117183

2412-0758/University of Technology-Iraq, Baghdad, Iraq This is an open access article under the CC BY 4.0 license http://creativecommons.org/licenses/by/4.0

INTRODUCTION

Ver the last few years, interest in tungsten trioxide (WO₃) has increased rapidly and significantly due to the material's potential applications in photo voltaic and photo catalytic processes^[1].Tungsten trioxide (WO₃) is a cheap material , with excellent chemical stability, nontoxicity, good mechanical properties and is one of the most efficient semiconductor photo catalyst for extensive environmental applications because of its strong oxidizing power, high photochemical corrosive resistance and cost effectiveness^[2,3] However, anatase is a wide band gap semiconductor (3.2-3.5 ev)^[7,8]. Anatase possesses a higher photocatalytic activity than rutile due to the difference in the optical band gap^[10].There are many techniques to synthesize WO₃ thin films, including sol–gel ,sputtering , anodic oxidation, pulsed laser deposition (PLD) , electron-beam evaporation and spray pyrolysis ^[8]. Among available techniques is pulsed laser deposition (PLD) which is a cheap deposition procedure, allowing the growth of rough-surface films at atmospheric pressure and on large area [11].

PLD plays a great roll in reducing the chemical contamination due to: the use of laser light, controlling of the composition of deposited structure, and in situ doping. Moreover, it is a versatile and powerful tool for production of nanoparticles with desired size and composition by only manipulating the deposition conditions. [3,4] It has been always applied as a gas sensing material to detect combustible, toxic and pollutant gases due to its high sensitivity, simple design and its low cost[5].

We report here the deposition of Tungsten trioxide thin film substrate at oxygen pressure of $(2 \times 10^{-1} \text{ mbar})$ and 1.2 J/cm² laser fluence and different substrate temperature. We also investigated the influence of different substrate temperature on structural ,morphological and sensing properties of the films.

Experimental Work

The deposition was carried out using a Q switched Nd:YAG laser at 532 nm (pulse width 7 n sec and laser fluence 1.2 J/cm²). Nd:YAG laser (HuafeiTongdaTechnology—DIAMOND-288 pattern EPLS) was used for the deposition of WO₃ on different substrates . Tungsten trioxide powder was taken from Fluka Company with high purity (99.99%) and pressed under a 10 tons to form a pellet with 2.5 cm diameter and 0.4 cm thickness. X-ray diffraction measurements (Philips PW 1050 X-ray diffractometer) have been conducted according the ASTM (American Society of Testing Materials) cards, using 1.54 Å from Cu-k_a. In order to study the surface morphology atomic force microscopy (AFM) (Digital Instruments Nanoscope II). A double–beam UV-VIS (210A Spectrophotometer) was used to measure the transmittance of WO₃ film deposited under different

conditions. Finally, film thickness was measured by using an optical interferometer and found to be around 153 nm.

Results and discussion

The X-ray diffraction of TrioxideWO₃Thin Film:

WO₃ films formed at substrate temperatures of $(25^{\circ}C, 300^{\circ}C, 400^{\circ}C)$ on glass substrate with a laser fluence of 1.2 J/cm^2 and Oxygen pressure of 2×10^{-1} mbar were used . Fig (1) shows the XRD measurements results for different peaks on $2\theta=23^{\circ},20^{\circ}$

=29.3765°, 2θ =33.02°, 2θ =50°, 2θ =53.75° and 2θ =54.11° corresponding to the (001), (101), (111), (102), (112) peaks respectively with low intensity. Fig (1) shows dominant peaks on 2θ =23.652°.



Figure (1):- XRD patterns of WO₃ thin films deposited at Oxygen pressure of 2 ×10⁻¹ mbar (a) RT(b) 300 °C, (c)400 °C.

Fig (2) shows The doped tungsten trioxide films become less crystalline than undoped samples. The peaks in the X- Ray diffraction shift into the region of higher θ , indicating stress in the grains. XRD analysis also did not detect the dopant phase ^[13].



Figure (2):- XRD patterns of WO₃:PT thin films deposited at Oxygen pressure of 2 ×10⁻¹mbar and at different concentrations a) 1% b) 3%, c)5%.

Atomic Force Microscopy (AFM)

Figure (3) shows the granularity accumulation distribution chart of WO₃ for WO₃ pure and doped with Pt films deposited at temperature of 400°C and on glass substrate by using PLD technique .The average grain size was found to be 115-140 nm from AFM. The average grain size was obtained by using Scherrers- Debye formula^[13] it was smaller than that estimated from AFM measurement indicating that grains are probably an aggregate of many crystallites.



doped Pt at 5%

From Fig (4) shows the optical transmittance of the undoped WO_3 films deposited on glass substrate at different temperatures. From all the films analyzed it is observed that the optical transmittance increases with increasing the temperatures.



Figure (4): Transmittance spectra of WO₃ thin films at different temperatures with fixed Oxygen pressure of 2 ×10⁻¹mbar

From Fig (5a,b,c,d) shows The direct band gap values for WO₃:Pt thin film doping at different Pt concentrations(1,3,5)% at substrate temperature of 400°C. The results show an increase in the Pt content resulted in a decrease in the band gap to about (3.8-3.5) eV. This decrease in energy gap can be due to the prohibited impurities that led to the formation of donor levels within the energy gap near the conduction band



Figure(5): A plot of $(\alpha h \upsilon)^2$ versus (h υ) of Wo₃ thin films at400 °C 400 at different doping concentrations with Pt a- as deposited, b-WO₃ at (1%) , c- WO₃:pt (3%) and d- WO₃:pt (5%)

The Electrical Properties

Figure (6) shows the resistivity decreases as the temperature is increased, with different thicknesses. This agrees with semiconductor behavior.



Figure (6) Ln σ versus 1000/T for WO₃ films at different thicknesses (a): 135nm, (b): 145nm, (c): 153nm, with different Substrate temperatures.

Table (1): D.C conductivity parameters	s for WO ₃ films at different thicknesses and
temp	peratures.

Thickness(nm)	T _a (K)	(303 – 363)K	(363 – 433)K	(403 – 483)K	$\sigma_{R,T} \times 10^{-3}$
		E _{a1} (eV)	$E_{a2}(eV)$	E _{a3} (eV)	(Ω.cm) ⁻¹
135	R.T	0.069	0.159	0.346	0.115
	373	0.075	0.179	0.376	0.077
	473	0.089	0.215	0.411	0.062
145	R.T	0.055	0.143	0.335	0.314
	373	0.064	0.161	0.361	0. 208
	473	0.0755	0.172	0.386	0.182
153	RT	0.0466	0.1	0.321	0.894
	373	0.061	0.125	0.332	0.577
	473	0.067	0.132	0.358	0.496

electrical resistivity was found to be a function of doping concentration as shown in Fig (7).

Eng. & Tech. Journal, Vol. 33, Part (B), No. 8, 2015



Figure (7): The electrical resistivity as a function of different doping concentrations with Platinum

Figure (8) shows that the electrical resistivity decreases with increasing doping concentration of Pt in WO_3 thin films. The results may be attributed to the increasing concentration (n).



Figure (8): Ln ρ as a function of 1000/T(K) ⁻¹ for WO3 films at different doping concentrations with Platinum(wo₃:1%pt,wo₃:3%pt wo₃:5%pt)

Table (2): Activation energies Ea₁ and Ea₂ for WO3₂ thin films for different doping concentrations with Platinum(wo₃:1%pt,wo₃:3%pt wo₃:5%pt)

Doping with Cr	E _{a1} (ev)	E _{a2} (ev)
wo3:1%pt	0.0957	0.11388
wo3:3%pt	0.10383	0.1175
wo 3:5%pt	0.1279	0.13383

Eng. & Tech. Journal, Vol. 33, Part (B), No. 8, 2015





Figure (9): sensitivity of WO₃ films to NO₂ gas at different operation temperatures (RT,300,400) ⁰C at laser fluence of 1.2 J/cm².

Fig (9) shows the sensitivity as a function of operation temperature in the range of (50-400 °C) for WO₃ pure and doped with noble metal at different concentrations of (1 , 3 and 5%). All the films increases with the increasing in the operating temperature ,reaching a maximum value corresponding to an optimum operating temperature which is 350° C for all the samples. The response of the undoped sensor to NO₂ gas is relatively low.



Figure (10): Sensitivity of WO₃: Pt films to NO₂ gas at different operation temperatures at laser fluence of 1.2 J/cm².

Fig (10) shows Noble metal doping (Pt) increases the sensitivity of WO₃:PT sensors to NO₂ gas and improves the sensor response at which the sensor response is maximized at 250° C for WO₃ doping 5% Pt.

CONCLUSION

From this work ,it can be concluded that best conditions of WO₃ pure and doping with noble metal for gas sensing prepared by PLD the substrate temperature deposition is 400 °C. The oxygen pressure is 2×10^{-1} mbar. The laser fluence energy density is 1.2 J/cm². The doping at percent (5 wt %)

Thus for sensing performance of WO₃ pure and dopant with noble metals modified sensors; it is clear that pure WO₃ showed poor response to NO₂ gas. 5 % wt Pt doped WO₃ thin film was the most sensitive element to NO₂ gas. The optimum operating temperature for NO₂ gas sensing was (250) $^{\circ}$ C

Pt doped WO₃ thin film would be suitable for fabricating the NO₂ gas sensors showed good selectivity to NO₂ gas.

REFERENCES

[1] Fumiaki Mitsugi ,Eiichi Hiraiwa, Tomoaki Ikegami, Kenji Ebihara and Raj Kumar Thareja "WO₃ thin films prepared by pulsed laser deposition", Japanese journal of applied physics, Vol. 41, P.P. 5372–5375, (2002).

[2] S. K. Gullapalli, R. S. Vemuri, and C. V. Ramana "Structural transformation induced changes in the optical properties of nanocrystalline tungsten oxide thin films", Applied physics letters, Vol. 96, P.P. 1–3, (2010).

[3]K.J. Patel, CJ Panchal, VA Kheraj, MS Desai"Growth, structural, electrical and optical properties of the thermally evaporated tungsten trioxide (WO ₃) thin films", Materials chemistry and science, Vol. 114, Issue 1, P.P. 475–478, (2009).

[4] E.Gyorgy, G.Socol and I. N. Mihailescu " Structural and optical characterization of WO₃ thin films for gas sensing applications", Journal of applied physics, Vol. 97, (2005).

[3] K. M. Karuppasamy "Studies on the electrochromic and photocatalytic properties of pure and vanadium doped tungsten oxide thin films prepared by electron beam evaporation and DC magnetron sputtering techniques" Ph. D thesis, Indian institute of technology madras channel, Department of Physics, (2008).

[4] Rao M.C.and Hussain O.M. "Optical properties of vacuum evaporated WO₃ thin films", Journal of chemical sciences, Vol. 1, No. 7, P.P. 76–80, (2011).

[5] George. F. Fine, Leon M. Cavanagh, Ayo Afonja and Russell Binions "Metal oxide semi-conductor gas sensors environmental monitoring", Sensors, Vol. 10, P.P. 5469–5502, (2010).

[6] M. C. Rao "Effect of substrate temperature on the structural and electrical conduction behavior of vacuum evaporated WO₃ thin film, Journal of optoelectronics and biomedical materials, Vol. 3, Issue 2, P. P. 45–50, (2011).

[7] <u>Rougier, A; Portemer, F; Quede, A; El Marssi, M</u>"Characterization of pulsed laser deposited WO₃ thin films for electrochromic devices", Journal of Applied Surface Science, Vol. 153, P.P. 1–9, (1999).

[8] Maosong Tong, Guorui Dai, Yuanda Wu, Xiuli He, Dingsan Gao"WO₃ thin film prepared by PECVD technique and its gas sensing properties to NO₂", Journal of materials science, Vol. 36, P.P. 2535–2538. (2001).

[9]Hiroharu Kawasaki, Jun Nuba, Keitarou Iwatsuji, Yoshiak Suda "NOx gas sensing properties of tungsten oxide thin films synthesized by pulsed laser deposition method", Journal of applied surface science, Vol. 197, P.P. 547–551, (2002).

[10]N.E. Stankova, G. Ferrante, and M. Zarcone, "Thin (0 0 1) tungsten trioxide films grown by laser deposition ", Journal of applied surface science, Vol. 247, P.P. 401–405, (2005).

[11]C. V. Ramana, Gaurav Baghmar , Ernesto J. Rubio , and Manuel J"Structural stability and phase transitions in WO_3 thin films", J. Phys. Chem. B, Vol. 110, P.P 10430–10435, (2006).

[12]Md. Mosharraf Hossain Bhuiyan , , Tsuyoshi Ueda, Tomoaki Ikegami and Kenji Ebihara "Gas sensing properties of metal doped WO_3 thin film sensors prepared by pulsed laser deposition and DC sputtering process", Japanese journal of applied physics, Vol. 45, P.P. 8469-8472, (2006).

[13]J. Gaury, E.M Kelder, Eugene Bychkov "Characterization of Nb-doped WO₃ thin films produced by electrostatic spray deposition", Thin solid films, P.P. 32–39 (2013).

[14]K.I. Gnanasekar, E.Prebhu, V.Jayarman, T.Gnanaskrougn "Sensor grade nanostructured thin films of multicomponent semiconducting oxide materials by pulsed laser deposition", Advanced materials letters, Vol. 4, No. 6, P.P. 464–475, (2013).